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A comparative study of biodegradation of vinyl acetate by environmental strains

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Abstract Four Gram-negative strains, E3 2001, EC1 2004, EC3 3502 and EC2 3502, previously isolated from soil samples, were subjected to comparative studies in order to select the best vinyl acetate degrader for waste gas treatment. Comparison of biochemical and physiological tests as well as the results of fatty acids analyses were comparable with the results of 16S rRNA gene sequence analyses. The isolated strains were identified as Pseudomonas putida EC3 2001, Pseudomonas putida EC1 2004, Achromobacter xylosoxidans EC3 3502 and Agrobacterium sp. EC2 3502 strains. Two additional strains, Pseudomonas fluorescens PCM 2123 and Stenotrophomonas malthophilia KB2, were used as controls. All described strains were able to use vinyl acetate as the only source of carbon and energy under aerobic as well as oxygen deficiency conditions. Esterase, alcohol dehydrogenase and aldehyde dehydrogenase were involved in vinyl acetate decomposition under aerobic conditions. Shorter degradation times of vinyl acetate were associated with accumulation of acetic acid, acetaldehyde and ethanol as intermediates in the culture fluids of EC3 2001 and KB2 strains. Complete aerobic degradation of vinyl acetate combined with a low increase in biomass was observed for EC3 2001 and EC1 2004 strains. In conclusion, P. putida EC1 2004 is proposed as the best vinyl acetate degrader for future waste gas treatment in trickle-bed bioreactors.

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Introduction

Numerous methods are used in the removal of different volatile organic compounds (VOCs) from gaseous streams (Kennes and Veiga 2001). Nowadays, biofiltration is thought to be the most cost-effective, reliable and environmentally friendly technology (Malhautier et al. 2005). In general, biofiltration is based on the transfer of pollutants from the air to the water phase, and subsequent decomposition by microorganisms to nontoxic metabolic end products, carbon dioxide and water as well as to biomass. Unfortunately, the formation of an excessive amount of biomass can sometimes take place, resulting in clogging of the bioreactor (Iliuta and Larachi 2004, 2005; Mendoza et al. 2004). Furthermore, decomposition of contaminants does not always lead to harmless products. Degradation of vinyl acetate goes through hydrolysis of the ester bond yielding acetic acid and vinyl alcohol. The latter quickly undergoes tautomerism to acetaldehyde (Nieder et al. 1990). Acetaldehyde reacts directly with DNA, causing formation of DNA-protein crosslinks and chromosomal aberrations in eukaryotic cells. The toxicity of acetaldehyde as well as other compounds containing the aldehyde group can be so high that reduction or even inhibition of growth of bacterial cells from the genus Pseudomonas was observed (Jakoby and Narrod 1958). A very simple mechanism of acetaldehyde toxicity reduction using strain V2 was described by Nieder et al. (1990). This strain posses oxidoreductase [EC 1.1.1.-] which, in the presence of NADH, reduces acetaldehyde to its alcohol form; the alcohol is then oxidised in the presence of NAD⁺. In this way, periodic increases in the concentration of toxic acetaldehyde can be controlled temporarily via conversion to ethanol.



Selection of the bacterial strain seems to be the crucial step in the preparation of well-operating trickle-bed reactors in waste gas treatment technologies. In a previous paper, Greń et al. (2009) described the isolation of 41 morphologically distinct isolates able to grow in the presence of vinyl acetate. Four of these were found to grow continuously while vinyl acetate at a concentration of 400 g m⁻³ was applied every 24 h. The aim of the present research was to carry out a comparative study on those four bacterial strains in order to chose the best vinyl acetate degrader. In addition the laboratory strain Pseudomonas fluorescens PCM 2123, and the environmental strain Stenotrophomonas malthophilia KB2 (Guzik et al. 2009) were used as controls. The isolated strains were described and analysed, and the dynamics of vinyl acetate degradation under aerobic and anaerobic conditions were verified. Activities of enzymes crucial for vinyl acetate decomposition were measured in crude cell extracts after induction. It is worth noting that vinyl acetate is used commonly nowadays for production of polyvinyl acetate, polyvinyl alcohol and other polymers that are exploited finally in the manufacture of building materials, printing inks, plastics, lacquers and paints. Despite the fact that vinyl acetate is recognised to be a widespread environment pollution (US EPA 2007), only Nieder et al. (1990) and Hatanaka et al. (1989) have described bacterial strains involved in vinyl acetate decomposition.

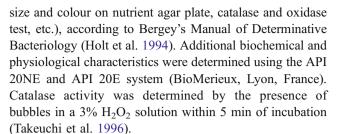
Materials and methods

Microorganisms

Strains: EC3 2001, EC1 2004, EC3 3502 and EC2 3502 were isolated from soil samples collected from ground on the site of Synthos S.A. in Oświecim (Poland) (formerly Chemical Company Dwory S.A.) in August 2006 using an enrichment culture technique as described previously (Greń et al. 2009). Pseudomonas fluorescens strain PCM 2123 was purchased from The Polish Collection of Microorganisms, Institute of Immunology and Experimental Therapy in Wrocław. Stenotrophomonas maltophilia KB2 came from the collection of Department of Biochemistry, University of Silesia in Katowice, and was originally isolated from the activated sludge of a sewage treatment plant in Bytom Miechowice in Poland using the classical enrichment technique with phenol as a selection factor (Guzik et al. 2009). Strains were kept on nutrient agar slopes at 4°C and transferred to the new medium monthly.

Biochemical identification

The isolated strains were characterised phenotypically and biochemically using standard techniques (colony shape,



Isolation of fatty acids was performed according to Sasser (1990). Analyses of fatty acid methyl esters (FAMEs) were performed using an HP 5890 gas chromatograph (Hewlett Packard, Rolling Meadows, IL) equipped with an HP 25 m× 0.2 mm cross-linked methyl-silicone capillary column. The initial oven temperature was 170°C, increased 5°Cmin⁻¹ to 260°C, increased further by 40°Cmin⁻¹ to 320°C and held constant for 1.5 min (Glucksman et al. 2000). Helium was used as the carrier gas. FAMEs were identified with Sherlock software (TSBA library, version 3.9, Microbial ID, Newark, NJ), based on the calibration of retention times of standards run prior to sample analysis.

Molecular identification

Bacterial DNAs were isolated from pure cultures using the DNA Mini Prep Kit (Qiagen). For 16S rRNA gene amplification the bacteria-specific primers 8F 5'AGTTTGAT CATCGCTCAG 3' and 1492R 5'GGTTACCTTGTTAC GACTT3' were used (Lonergan et al. 1996). Amplifications were carried out through a program consisting of initial denaturation at 94°C for 300 s, 3 cycles at 94°C for 45 s, 57°C for 30 s, 72°C for 120 s; 3 cycles at 94°C for 45 s, 56°C for 30 s, 72°C for 120 s; 3 cycles at 94°C for 45 s, 55°C for 30 s, 72°C for 120 s; 26 cycles at 94°C for 45 s, 53°C for 30 s, 72°C for 120 s; and a final elongation cycle at 72°C for 300 s. Nucleotide sequencing of the genes was performed using a Big Dye^R Terminator Cycle Sequencing Kit (Applied Biosystems, Foster City, CA) and AbiPrism®3100 Genetic Analyzer. The MegaBLAST program was used for homology searches with the standard program defaults. Multiple sequence alignments were performed and the neighbourjoining phylogenetic tree was constructed using the CLC Free Workbench 4.5.1 program. The 16S rRNA gene sequences determined in this research have been deposited in the GenBank database of NCBI under the accession numbers: EU877075, EU877078, EU877076 and EU877077 for strains, EC3 2001, EC1 2004, EC3 3502 and EC2 3502, respectively.

Culture conditions

The research was conducted in a batch bioreactor with a functional capacity of 2 dm³. Bacteria were grown in mineral medium, the composition of which was described



previously (Greń et al. 2009), supplemented with 47 g m⁻³ or 124 g m⁻³ vinyl acetate. Every experiment was conducted under the same conditions: volume of the suspended matter 1,500 ml; pH 7; temperature 30°C; stirring 130 rpm. The pH of the medium was adjusted to 7 by adding a 10% solution of KOH. Because of the volatility of the substrate and intermediates oxygen could not be supplied by air sparging. A constant DO (dissolved oxygen) concentration (~5 mg dm⁻³) was achieved by dosage with a 0.75% solution of hydrogen peroxide.

Analytical methods

The composition of the fluid cultures was determined by means of gas chromatography. Samples were taken from the cultures under sterile conditions and analysed directly by injection of 0.15 µl samples into a Varian 3800 gas chromatograph, equipped with a 30 m length, 0.53 mm diameter CP-wax column and a flame ionisation detector (FID). Helium was used as the carrier gas. Separation was achieved with a temperature programme (30°C min⁻¹ from 70°C to 150°C). The temperature of injector and the temperature of the FID was 250°C.

The concentration of biomass was determined by measuring the absorbance of the fluid culture (λ = 550 nm).

Preparation of inoculum

The bacterial strains were grown in 100 ml mineral salts medium supplemented with 400 g m⁻³ vinyl acetate introduced directly to the medium every 24 h. The flasks were incubated with shaking (130 rpm) at 30°C. After 3 weeks of preincubation the cells were used as inoculum for the bioreactor.

Preparation of crude enzyme extract

Bacteria were pre-grown in 500 ml mineral medium supplemented with vinyl acetate at a concentration of 400 g m^{-3} added to the culture every 24 h as a source of carbon and energy. The strain was grown aerobically for 5 days in a shaker at 130 rpm in 30°C in 1,000 ml flasks before harvesting. Cells were harvested in the late exponential phase of growth by centrifugation at 4,612 g, for 20 min at 4°C. The supernatant was left and kept at 4°C to the analysis. The cells were washed in 100 ml of 100 mM phosphate buffer (pH 7.5), and resuspended in the same buffer. Disruption of the cells was performed by sonicating six times for 15 s. Cell debris was removed by centrifugation (9,000 g, for 30 min at 4°C). The clear supernatant was used as the crude cell extract for enzyme assays.

Protein assays

The protein concentration was estimated using the Bradford method with lysozyme as a standard (Bradford 1976).

Enzyme assays

Esterase activity was determined spectrophotometrically using p-nitrophenyl butyrate (pNPB) dissolved in dimethyl sulfoxide (DMSO) as a substrate. Esterase activity was determined in 100 mM phosphate buffer (pH 7.5), and the final concentration of substrate in the reaction mixture was 1 mM. The amount of liberated p-nitrophenol (pNP) was determined at 410 nm at 30°C. The increase in absorbance was monitored for 3 min after addition of 25 μ l crude cell extract. The molar absorbance coefficient for pNPB was 18,400 M⁻¹ cm⁻¹. One unit (U) of esterase activity was defined as the amount of enzyme releasing 1 μ mol pNP per minute under assay conditions.

Dehydrogenase activities were determined spectrophotometrically in 100 mM phosphate buffer with 750 μ l 20 mM NAD⁺ or NADH and 100 μ l crude cell extract at 340 nm at room temperature. Reaction substrates were: 95% ethanol for alcohol dehydrogenase and 100 mM acetaldehyde for aldehyde and alcohol dehydrogenase. One unit of enzyme was defined as the amount of enzyme catalysing formation of 1 μ mol NADH/ NAD⁺ per minute. The molar absorbance coefficient for NADH/NAD⁺ at 340 nm was 6,220 M⁻¹ cm⁻¹.

Results and discussion

Identification of isolated strains degrading high concentrations of vinyl acetate

Identification of the four Gram-negative strains was based on morphological, biochemical and genetic methods. All strains grew on nutrient agar plates as circular, opaque, creamy-coloured colonies. The biochemical and physiological characteristics of the isolated strains are summarised in Table 1. Three of the isolated strains exhibited catalase and oxidase activities, while nitrate reduction was found only in cells of strains denoted as EC2_3502 and EC3_3502. Glucose was the only substrate assimilated by all isolated strains.

Analysis of the fatty acids pattern (Table 2) showed the largest contribution of fatty acids with a hydroxyl in the third position from the acid end and palmitic acid, which were listed by Kozdrój and van Elsas (2001) as most typical of Gram-negative bacteria and the genus *Pseudomonas*, respectively. The patterns generated with API 20NE tests were in good agreement with the results obtained from fatty acid patterns and allowed strains



Table 1 Differential phenotypic characteristics of isolated strains

Characteristic	EC3_2001	EC1_2004	EC2_3502	EC3_3502
Nitrate reduction	_	_	+	+
Indol production	_	_	_	_
Fermentation of glucose	=	_	=	_
Arginine dihydrolase	+	+	=	_
Urease	=	_	=	_
Hydrolysis of esculin	_	_	+	_
Hydrolysis of gelatin	_	_	_	_
Oxidase	+	+	_	+
Catalase	+	+	=	+
Beta-galactosidase	=	_	+	_
Assimilation of:				
Glucose	+	+	+	+
Arabinose	_	+	+	_
Mannose	+	_	+	_
Mannitol	_	+	+	_
N-acetyl-glucosamine	=	_	+	_
Maltose	=	_	+	_
Gluconate	+	_	=	+
Caprate	+	+	_	_
Adipate	_	+	_	_
Malate	+	=	+	+
Citrate	+	+	_	+
Phenylacetate	+	+	_	+

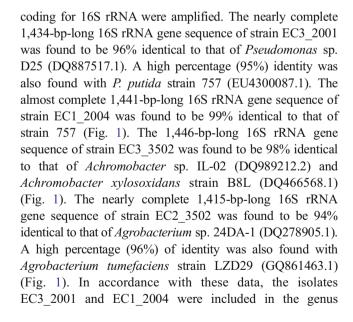
EC3_2001 and EC1_2004 to be classified as *Pseudomonas* putida, strain EC3_3502 as *Achromobacter xylosooxidans* and EC2_3502 as *Agrobacterium* sp.

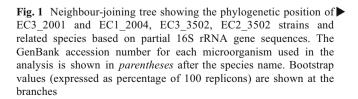
In order to confirm the phylogenetic relationships of the isolated strains, genomic DNAs were isolated and the genes

Table 2 Percentage of total fatty acids from the isolated strains

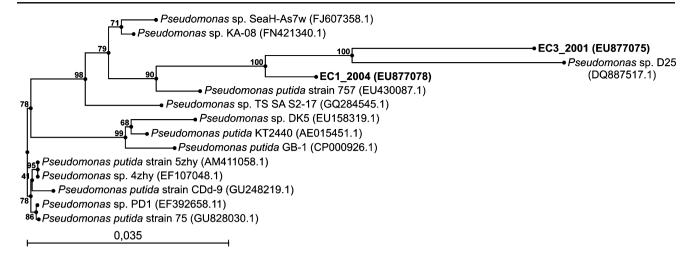
Fatty acid ^a	EC3_2001	EC1_2004	EC2_3502	EC3_3502
10:0 3OH	7.82	7.47	-	_
12:0	2.03	1.82	-	_
12:0 2OH	7.69	6.26	-	_
12:0 3OH	6.32	5.46	-	_
14:0	_	_	_	1.49
15:0 anteiso	_	1.18	_	=
16:0	27.79	26.61	8.61	38.19
16:0 3OH	_	_	4.98	_
17:0 anteiso	0.93	1.04	1.47	_
17:0 cyclo	22.03	11.06	-	11.24
18:0	_	-	-	3.17
19:0 <i>cyclo</i> ω 8c	1.33	_	4.39	_
19:0 10 methyl	_	=	1.15	_

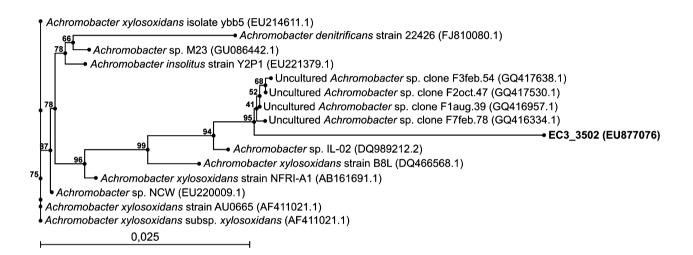
 $^{^{}a}$ ω Methyl end of fatty acid, c cis configuration of the double bond, cyclo cyclopropane fatty acid, iso branched fatty acids, OH indicates the position of hydroxyl group from the acid end

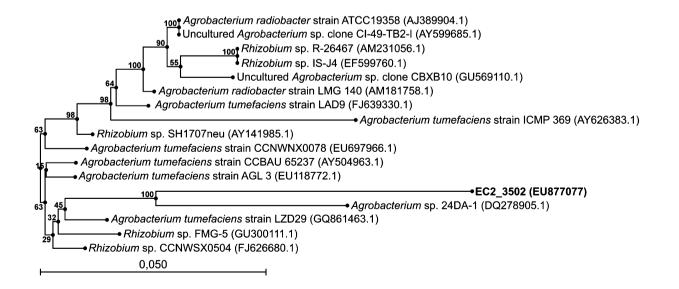














Pseudomonas and named as P. putida EC3_2001 and EC1_2004, respectively. The isolate EC3_3502 was included in the genus Achromobacter and EC2_3502 in the genus Agrobacterium and named as Achromobacter xylosoxidans EC3_3502 and Agrobacterium sp. EC2_3502, respectively.

Bacteria belonging to the genus *Pseudomonas* are known to be good degraders of numerous harmful substances such as aliphatic and aromatic hydrocarbons or fatty acids (Lalucat et al. 2006). Representatives of the genus *Achromobacter* are described as microorganisms able

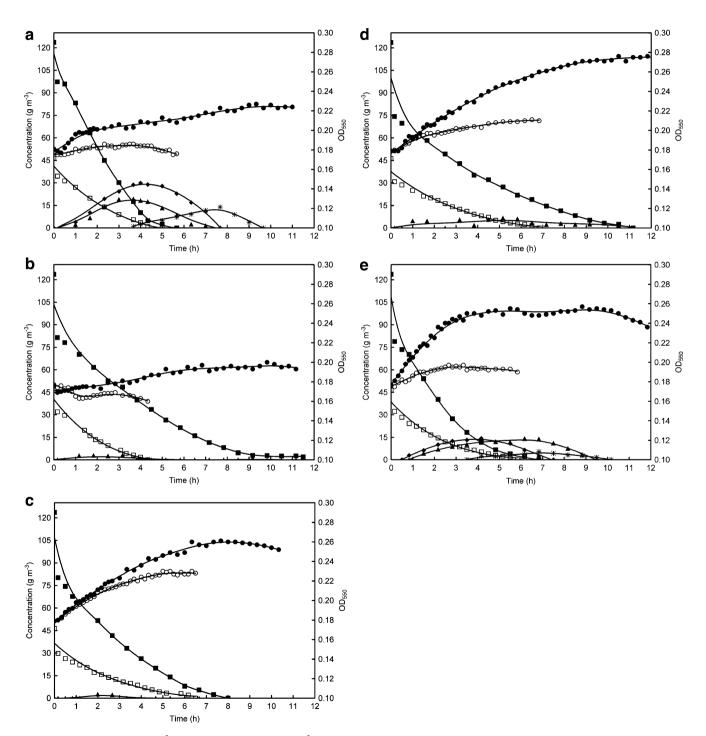


Fig. 2 Degradation of 47 g m⁻³ (*white symbols*) and 124 g m⁻³ (*black symbols*) of vinyl acetate under aerobic conditions. **a** EC3_2001, **b** EC1_2004, **c** EC3_3502, **d** PCM 2123, **e** KB2; *circles* optical

density at 550 nm, squares vinyl acetate, triangles acetic acid, diamonds acetaldehyde, asterisks ethanol



to degrade different chemical compounds including VOCs (Nielsen et al. 2006; Wan et al. 2007; Eixarch and Constanti 2010). There are some reports concerning decomposition of organic compounds by *Agrobacterium* strains, mainly *Agrobacterium radiobacter* (Higgins et al. 1993; Struthers et al. 1996). So far only Nieder et al. (1990) have described a Gram-negative strain (V2) using vinyl acetate as the sole source of carbon and energy, and Hatanaka et al. (1989) described esterases of *Pseudomonas* sp. Z2 strain, only one of which was involved in cleavage of the ester bond of vinyl acetate.

Degradation of vinyl acetate under aerobic conditions

Degradation of 47 g m⁻³ and 124 g m⁻³ vinyl acetate was studied for only three of the isolated species. Because *Agrobacterium* sp. strain EC2_3502 was catalase negative (Table 1), it was unable to decompose vinyl acetate in the presence of oxygen provided in hydrogen peroxide. As controls, *P. fluorescens* PCM 2123 and *Stenotrophomonas malthophilia* KB2 were used (Fig. 2).

The time required to degrade vinyl acetate at a concentration of 47 g m⁻³ varied between the described strains, with the shortest time being observed in cultures of P. putida strains. The fastest degradation rate of 124 g m⁻³ vinyl acetate was observed for strain P. putida EC3 2001, Achromobacter xylosoxidans EC3 3502 and Stenotrophomonas malthophilia KB2. But the result of quick hydrolysis of the ester bond was the accumulation of acetic acid and acetaldehyde in the culture fluid of EC3 2001 and KB2 strains. Acetaldehyde at a concentration of 10° g m⁻³ was found to retard microbial growth, and a dose of 13° g m⁻³ completely inhibited any increase in biomass but not vinyl acetate metabolism. A decrease in the acetaldehyde concentration was correlated with a temporary increase in ethanol concentration (Fig. 2a,e). A constant ethanol concentration of about 2–5 g m⁻³ was maintained during the entire incubation period in most cultures, but ethanol was finally metabolised directly to acetic acid because no successive accumulation of acetaldehyde was observed. In the cultures of three strains, EC3 3502, PCM

PCM 2123

KB2

 638 ± 166

 724 ± 40

 9.67 ± 3.13

 12.09 ± 4.55

Table 3 Specific esterase and dehydrogenases activities (mUmg⁻¹ protein) in crude cell extracts of the described strains after 5-days induction with vinyl acetate at a concentration of 400 g m⁻³ under aerobic conditions. The results are an average of at least three independent measurements

of vinyl acetate, in comparison to studies carried out under Alcohol dehydrogenase Alcohol dehydrogenase Esterase Aldehyde dehydrogenase Substrate pNPBEthanol Acetaldehyde Acetaldehyde Strain EC3 2001 694±46 10.22 ± 2.46 66.21 ± 12.22 7.21 ± 2.10 EC1 2004 323 ± 119 5.22 ± 1.45 3.11 ± 0.89 5.21 ± 2.12 EC2 3502 436 ± 120 11.34 ± 2.78 12.23 ± 3.29 23.66 ± 10.09 EC3 3502 720 ± 22 2.03 ± 0.15 10.02 ± 2.55 4.74 ± 1.34

 58.34 ± 16.17

 36.89 ± 10.02

2123 and KB2, complete transformation of vinyl acetate was seen together with proliferation of the bacterial strain (Fig. 2c–e). The same results were described by Nieder et al. (1990) for Gram-negative isolate V2. Intermediates of vinyl acetate hydrolysis did not appear in the culture fluid, and a visual increase in optical density of the culture accompanied the complete decomposition of vinyl acetate. In contrast, intensive growth of *Stenotrophomonas malthophilia* KB2 was observed until acetaldehyde appeared in the culture fluid, suggesting its toxic influence on cells of strain KB2.

Enzyme activities involved in vinyl acetate decomposition under aerobic conditions

The first step of vinyl acetate biodegradation by the described strains under aerobic conditions is hydrolysis of the ester bond due to esterase activities (Table 3). No esterase activities were found in the extracellular fraction obtained after centrifugation of the culture fluids of the described strains (data not shown). Higher esterase activity was combined with a higher vinyl acetate degradation rate and shorter decomposition time by EC3 2001, PCM 2123 and KB2 strains. The levels of oxidoreductases activities involved in the vinyl acetate degradation rate were lower in comparison to the esterase specific activities determined in crude cell extracts. The presence of both alcohol dehydrogenase and aldehyde dehydrogenase confirmed our previous suggestions that periodic transformation of acetaldehyde to ethanol could be a defence mechanism against acetaldehyde toxicity. Further studies on the kinetics of enzymes engaged in vinyl acetate degradation are ongoing.

Effect of oxygen deficiency on vinyl acetate decomposition

Due to the conditions of temporary oxygen shortage that occur during operation of a trickle-bed reactor, biodegradation of 124 g m⁻³ vinyl acetate under oxygen deficiency was studied for all isolated and control strains. Oxygen deficiency inhibited neither growth of bacterial strains nor degradation of vinyl acetate, in comparison to studies carried out under



 3.67 ± 1.23

 9.67 ± 1.11

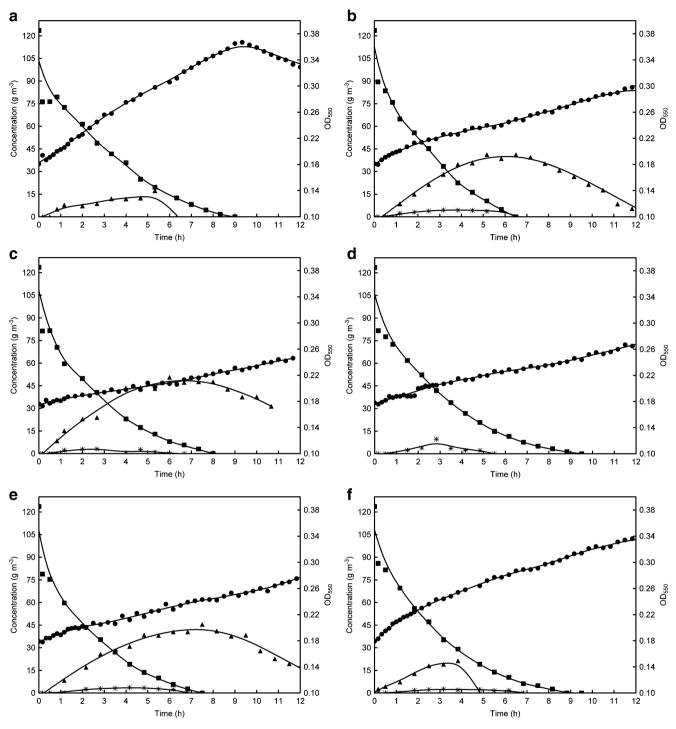


Fig. 3 Degradation of 124 g m⁻³ vinyl acetate under oxygen deficiency. a EC3_2001, b EC1_2004, c EC3_3502, d EC2_3502, e PCM 2123, f KB2; circles optical density at 550 nm, squares vinyl acetate, triangles acetic acid, diamonds acetaldehyde, asterisks ethanol

aerobic conditions (Fig. 3). The most abundant intermediate of vinyl acetate decomposition in all the described strains except for *Agrobacterium* sp. EC2_3502 (Fig. 3d) was acetic acid. The lowest concentrations of acetic acid were observed in the cultures of intensively proliferating cells of EC3_2001 and KB2 strains. In addition, small amounts of ethanol were detected in the culture fluids of all strains apart

from P. putida EC3_2001 (Fig. 3a). Ethanol was also found as an intermediate of incomplete decomposition of vinyl acetate under N_2 by strain V2 (Nieder et al. 1990), while the initial concentration of vinyl acetate did not exceed 15 mM. In the presence of higher concentrations of vinyl acetate, the presence of an additional intermediate—acetaldehyde—was confirmed.



Conclusions

Biochemical, physiological and genetic tests allowed the isolated strains to be identified as P. putida EC3 2001, P. putida EC1 2004, Achromobacter xylosoxidans EC3 3502 and Agrobacterium sp. EC2 3502. The described strains were able to effect the complete mineralisation of vinyl acetate without any accumulation of harmful end-products. Lack of catalase activity precluded usage of EC2 3502 strain in vinyl acetate decomposition under the required experimental conditions. The main criterion of strain selection was biomass production in order to prevent future biological clogging and physical plugging during operation of the trickle-bed bioreactor. Because the bioreactor operates for long periods under aerobic conditions, the two P. putida strains would seem to be the best microorganisms for vinyl acetate removal. Although the decomposition time of vinyl acetate by strain EC1 2004 was longer than that of EC3 2001, the absence of toxic intermediates during degradation makes P. putida EC1 2004 strain the best vinyl acetate degrader of the strains tested here.

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